

single-molecule fluorescence resonance energy transfer II

(6) homo-FRET

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25/02/2005

Energy transfer between identical fluorophores

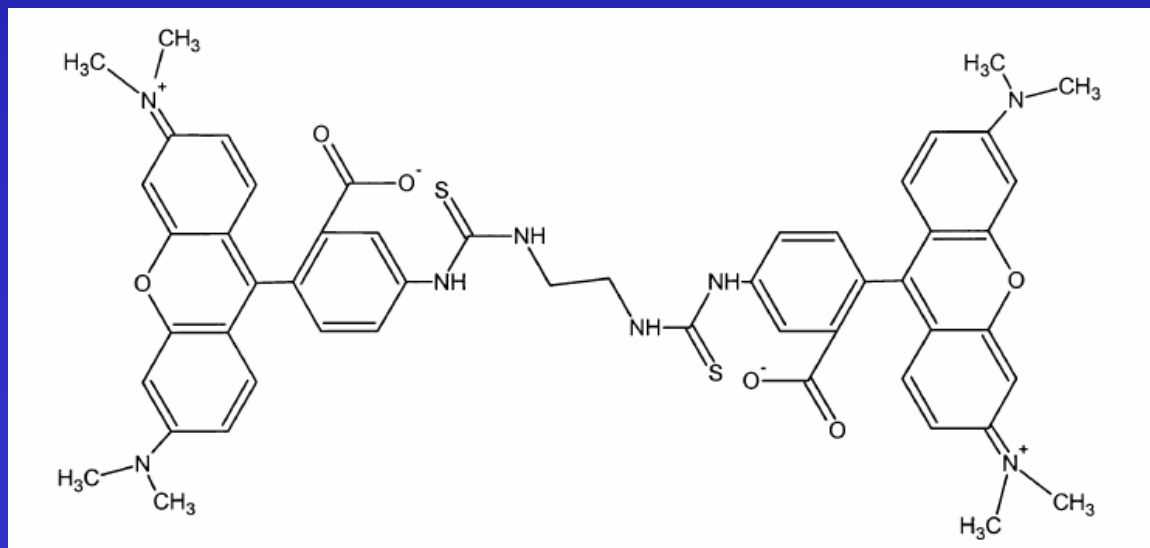
Examples

1. artificial homo-dimers
2. light harvesting complexes

Energy transfer between two identical fluorophores

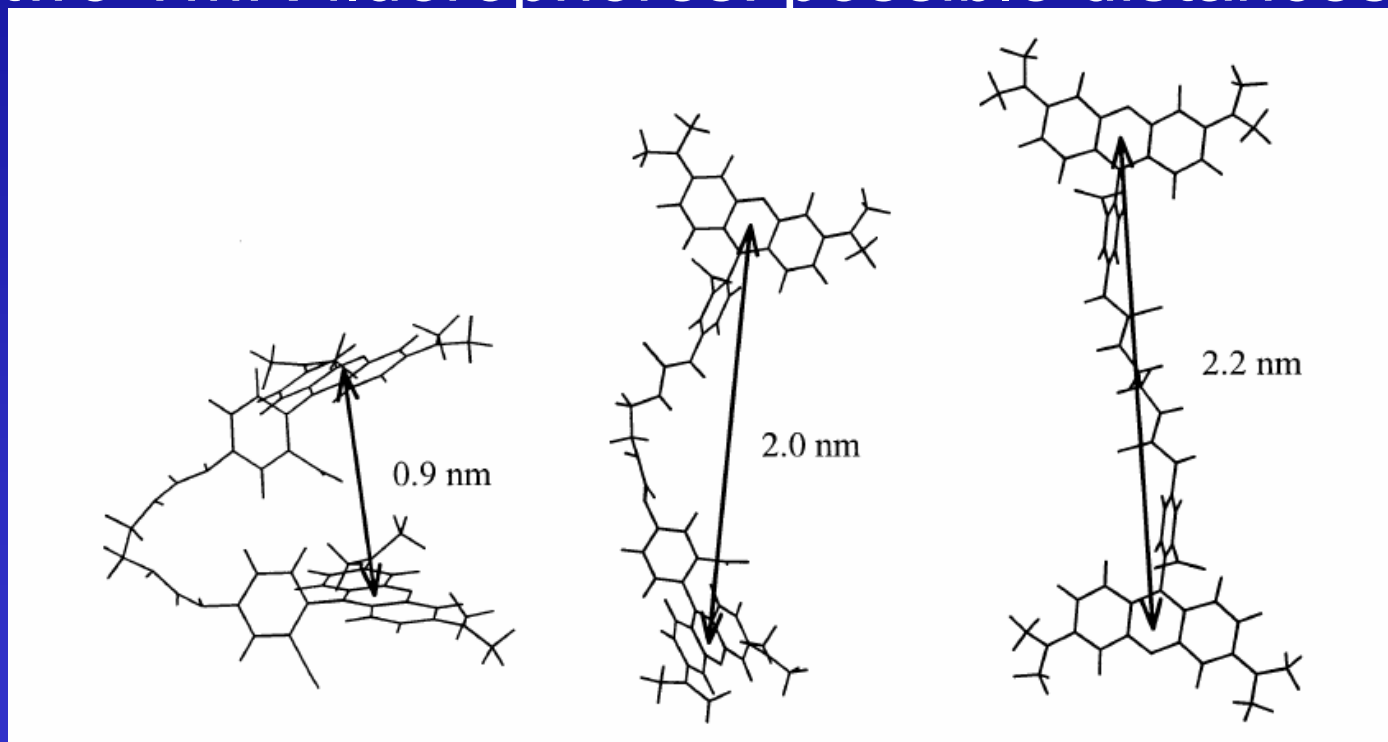
Examples

1a. artificial homo-dimers:
two crosslinked TMR-fluorophores "TRITC₂"



Energy transfer between two identical fluorophores

two TMR-fluorophores: possible distances



strong \leftrightarrow

weak coupling

Energy transfer between two identical fluorophores

two TMR-fluorophores ("TRITC₂): absorption spectra

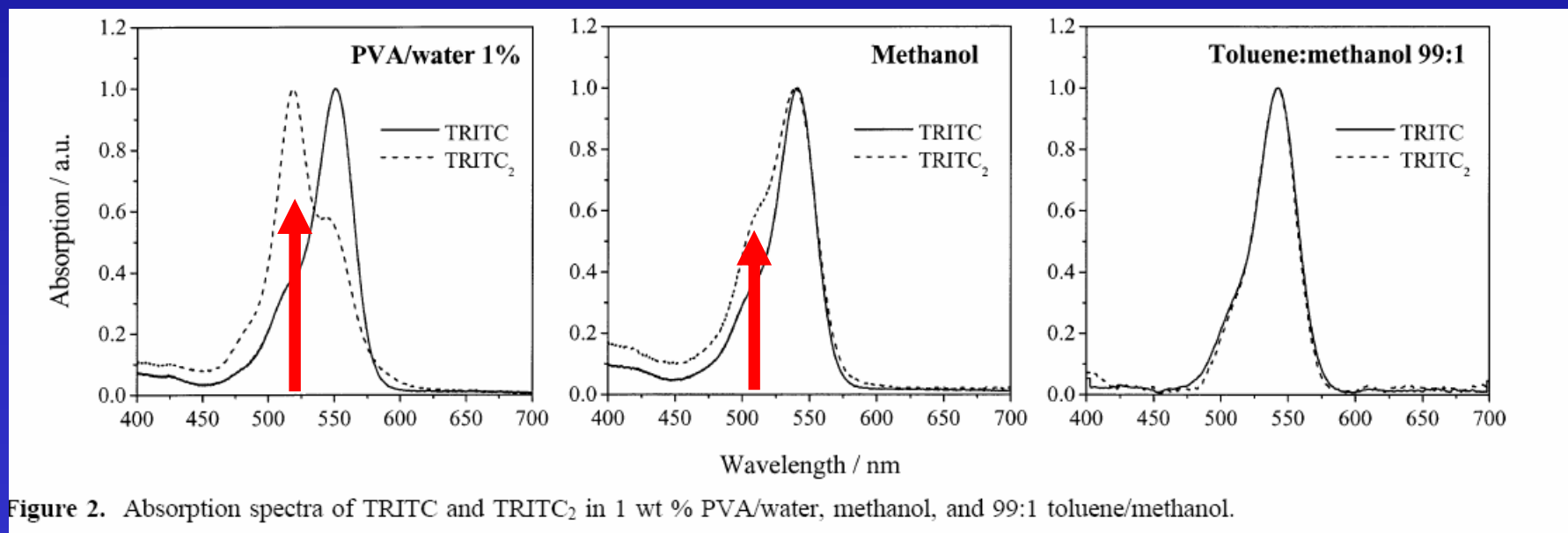


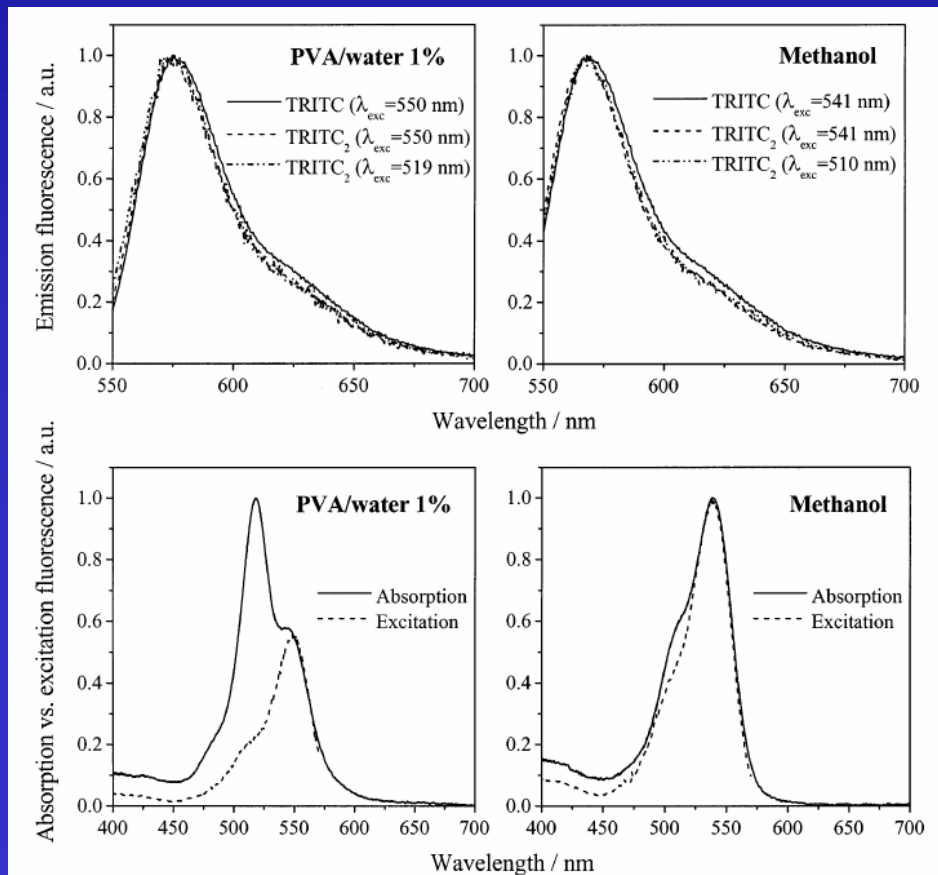
Figure 2. Absorption spectra of TRITC and TRITC₂ in 1 wt % PVA/water, methanol, and 99:1 toluene/methanol.

solvent dependent conformations of the dimer

Energy transfer between two identical fluorophores

two TMR-fluorophores ("TRITC₂): fluorescence shows

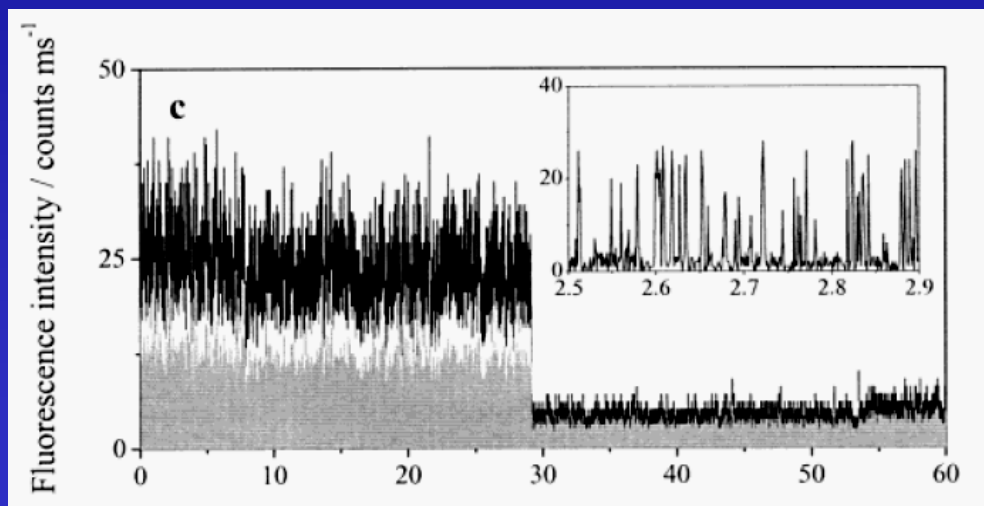
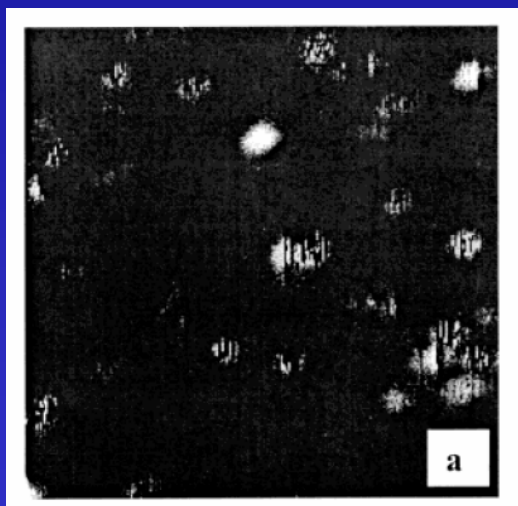
only one spectrum



excitation spectra →
a second population
with $\Phi_{FI}=0$ in water

Energy transfer between two identical fluorophores

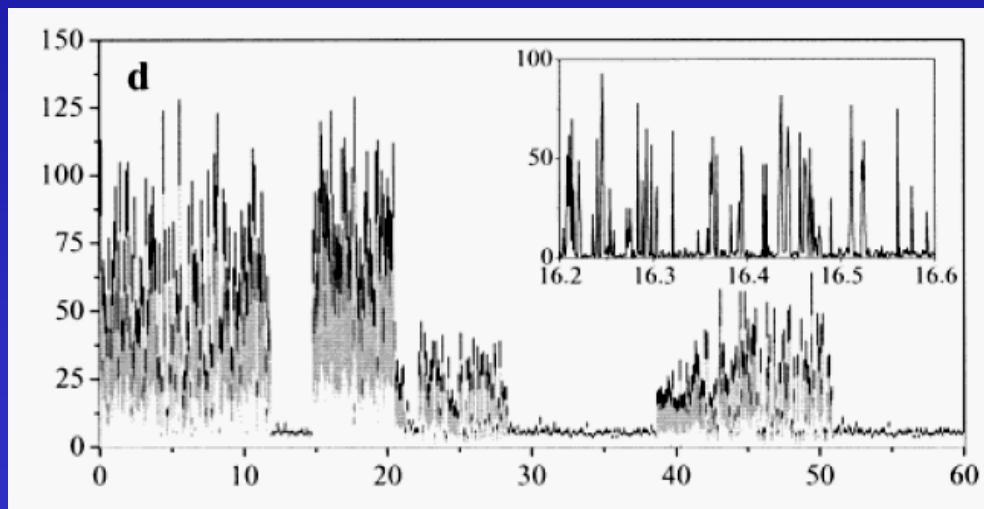
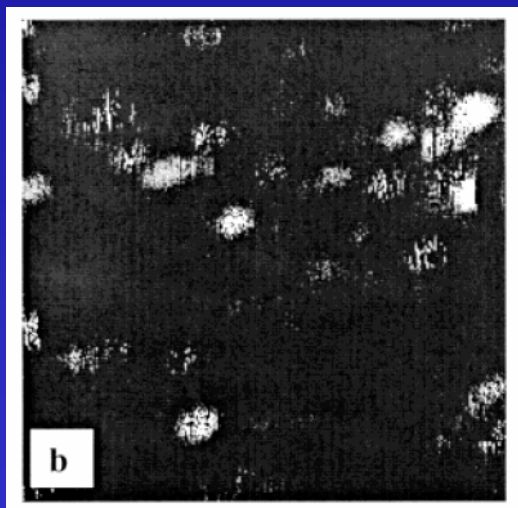
"TRITC": single monomer images



black: total fluorescence intensity
gray: two orthogonal polarizations
→ one-step photobleaching, constant intensity

Energy transfer between two identical fluorophores

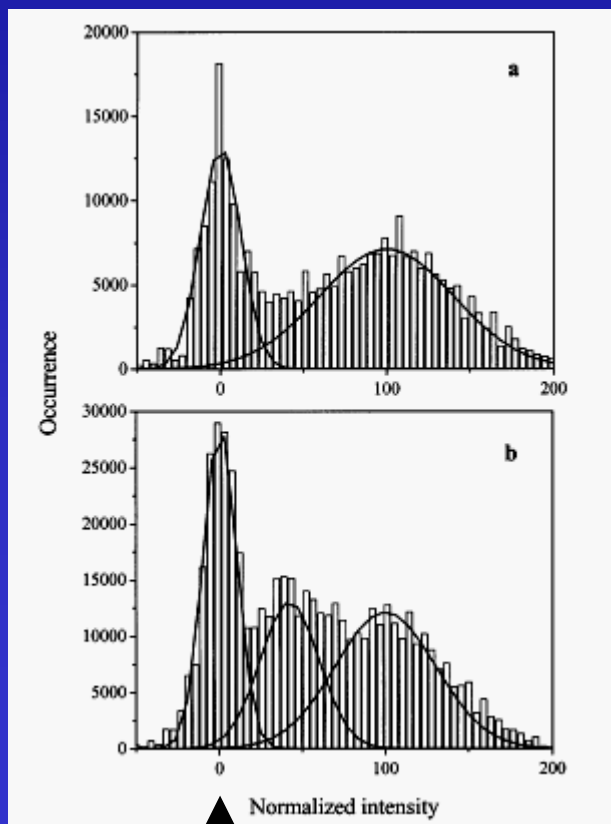
"TRITC₂": single-dimer images



black: total fluorescence intensity
gray: two orthogonal polarizations
→ two intensity levels

Energy transfer between two identical fluorophores

TRITC and TRITC₂: intensity distributions



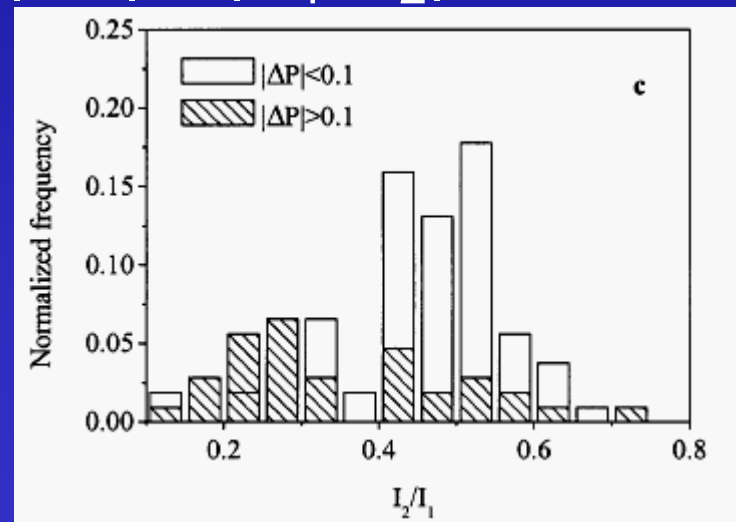
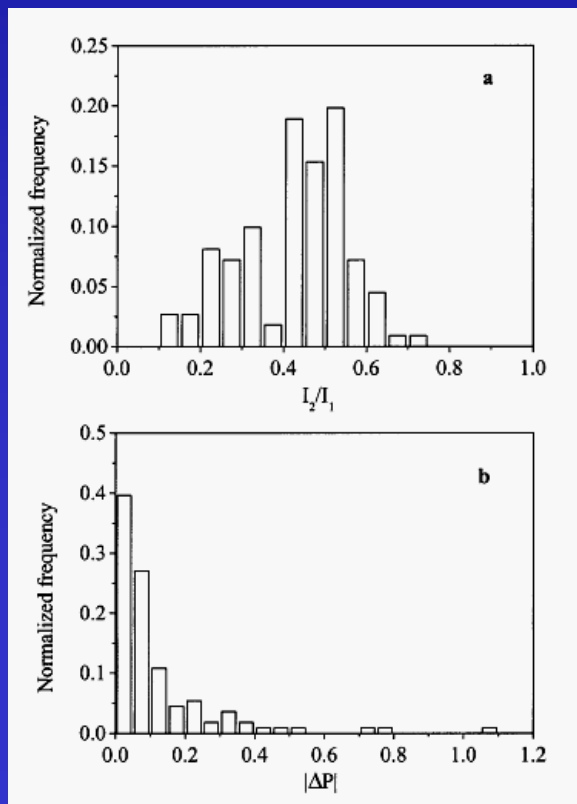
the monomer exhibits only one intensity level,

the dimer shows two distinct intensity levels I_1 and I_2 .

↑ background level

Energy transfer between two identical fluorophores

TRITC₂: intensity ratio I_2/I_1 and differences in linear polarization $|\Delta P| = |P_1 - P_2|$



small $|\Delta p|$ changes correspond to $I_2/I_1 = 0.5$ i.e. parallel transition dipoles in the dimer and the remaining monomer after bleaching

$$P = (I_s - I_p)/(I_s + I_p)$$

Energy transfer between two identical fluorophores

TRITC₂: two lifetimes (3.2 ns and 2.1 ns) and correlation with intensity levels:

3.2 ns ↔ higher intensity level

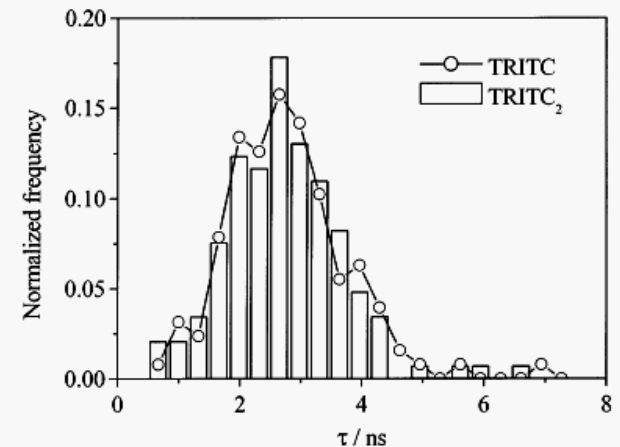
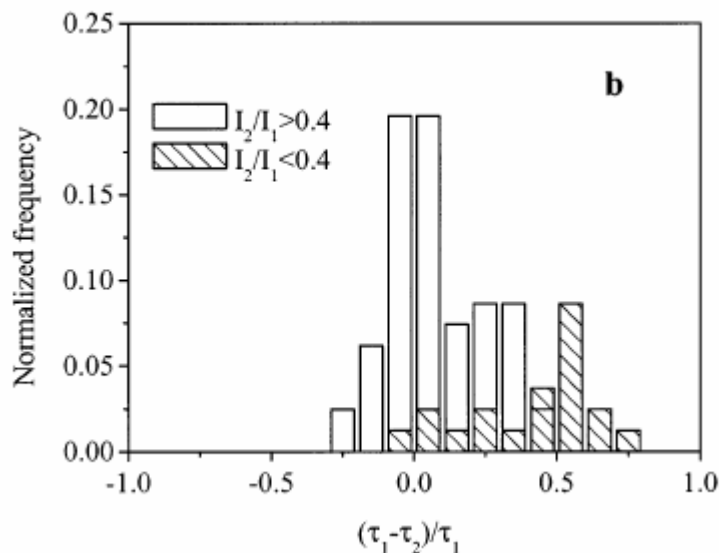


Figure 8. Distributions of fluorescence lifetimes for 127 TRITC and 146 TRITC₂ molecules ($\lambda_{\text{exc}} = 550 \text{ nm}$, power = 4.5 kW/cm²).

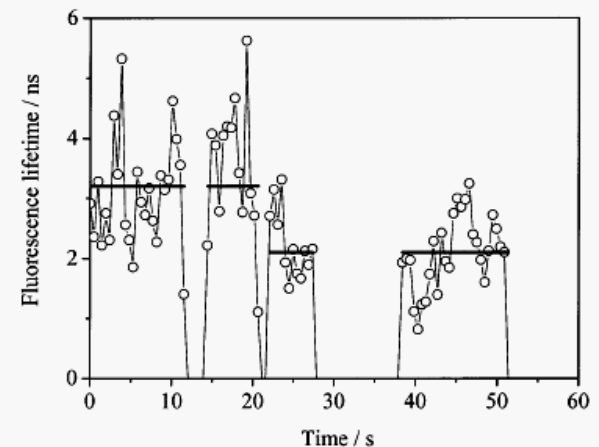


Figure 9. Fluorescence lifetime trajectory corresponding to the TRITC₂

Energy transfer between two identical fluorophores

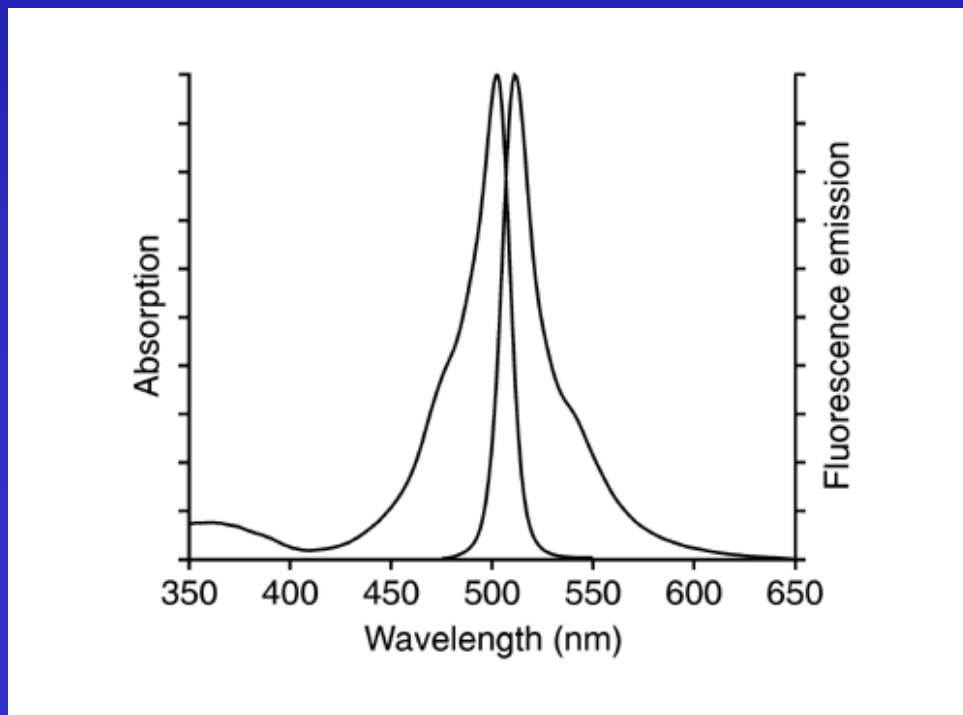
photophysics of TRITC₂ :

- strong coupling: non-fluorescent H-type dimer, concentration independent
- weak coupling: similar excitation and fluorescence spectra as monomer
- dimer emits more photons than monomer (6.5×10^5 vs. 3.4×10^5), shows two intensity levels (i.e. photobleaching of one dye before the second)
- small polarization changes Δp after photobleaching correlate with a 50% loss in intensity: this indicates parallel transition dipole moments in the dimer
- large polarization changes Δp after photobleaching correlate with $I_2/I_1 \neq 0.5$ i.e. non-parallel dipoles, as both dyes are not excited / detected with the same efficiency
- unimodal and similar lifetime distribution for monomer and dimer
- lifetime jumps: shorter lifetime for dimmer level indicates a remaining and absorbing chromophore after bleaching

Energy transfer between two identical fluorophores

Examples

1b. artificial homo-dimers in PAI 1



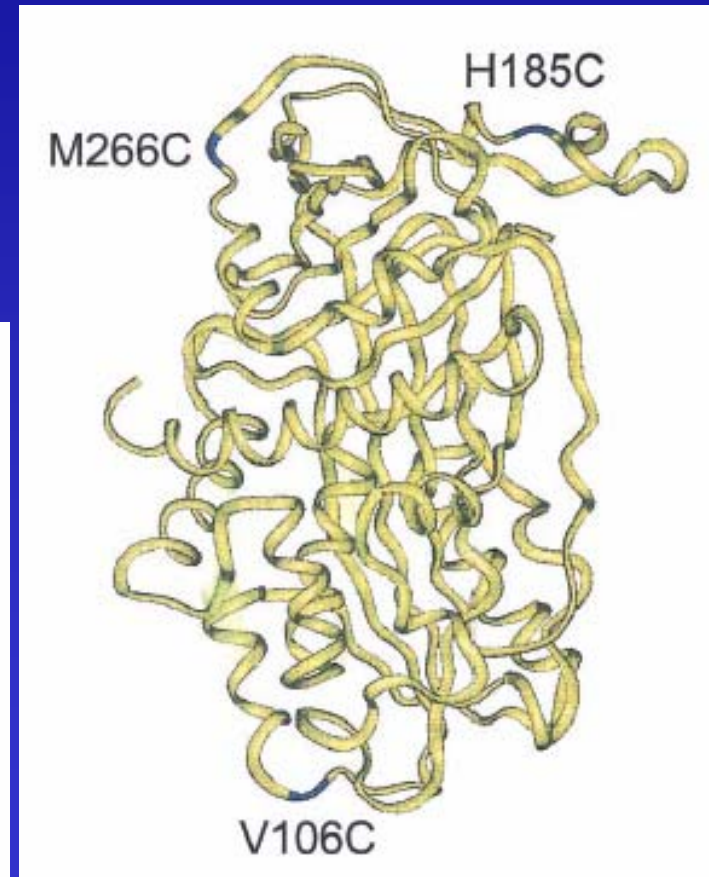
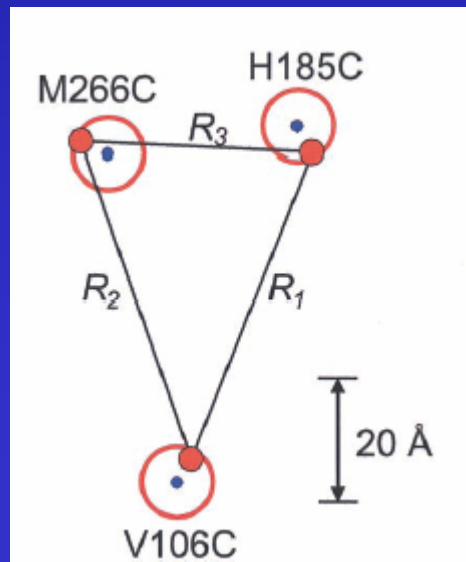
(plasminogen activator inhibitor type 1)

with two Bodipy-FL fluorophores

Energy transfer between two identical fluorophores

PAI1

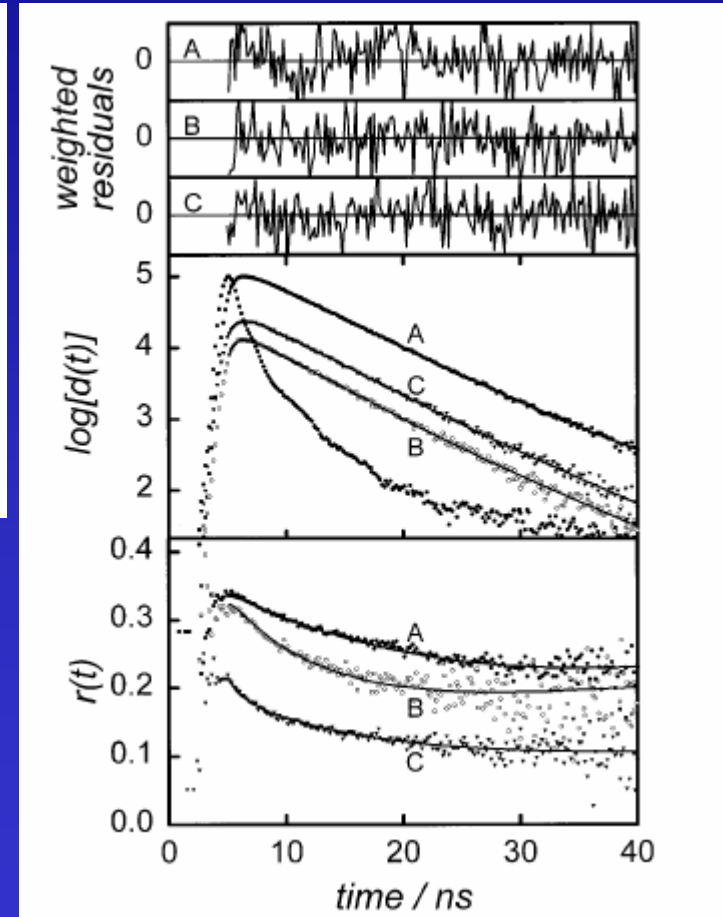
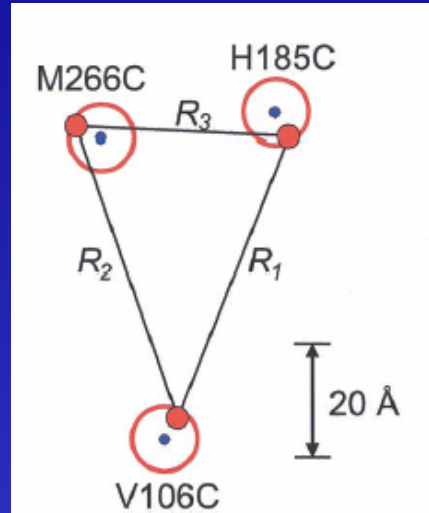
- three cysteines
- mono- and dual-labeled with SBDY
- timeresolved anisotropy measurements



Energy transfer between two identical fluorophores

PAI1

- three cysteines
 - A: single cys 185
 - B: single cys 266
 - C: double cys 185-266
-
- timeresolved anisotropy measurements
→ faster decay for double cys185-266



Energy transfer between two identical fluorophores

rate ω for donor-donor energy migration and fluorophore distance R :

$$\omega = \frac{3}{2} \langle \kappa^2 \rangle \frac{1}{\tau} \left\{ \frac{R_0}{R} \right\}^6$$

Mutant	r_s	$r(0)$	ϕ/ns	ρ_0	S_j	S_8	ω/ns^{-1}	$\langle \kappa^2 \rangle$	$R/\text{\AA}$	$R_c/\text{\AA}$
V106C	0.303	0.332	12 ± 1		0.80 ± 0.02					
H185C	0.306	0.343	12 ± 1		0.78 ± 0.02					
M266C	0.264	0.329	7 ± 1.5		0.69 ± 0.01					
V106C- H185C	0.272	0.340		0.81		0.17	0.6 ± 0.1	1.5	54 ± 4	60.9
H185C- M266C	0.167	0.329		-0.26		-0.36	2.1 ± 0.3	0.6	38 ± 3	30.8
M266C- V106C	0.22	0.333		0.40		0.06	0.4 ± 0.1	1.3	55 ± 3	55.1

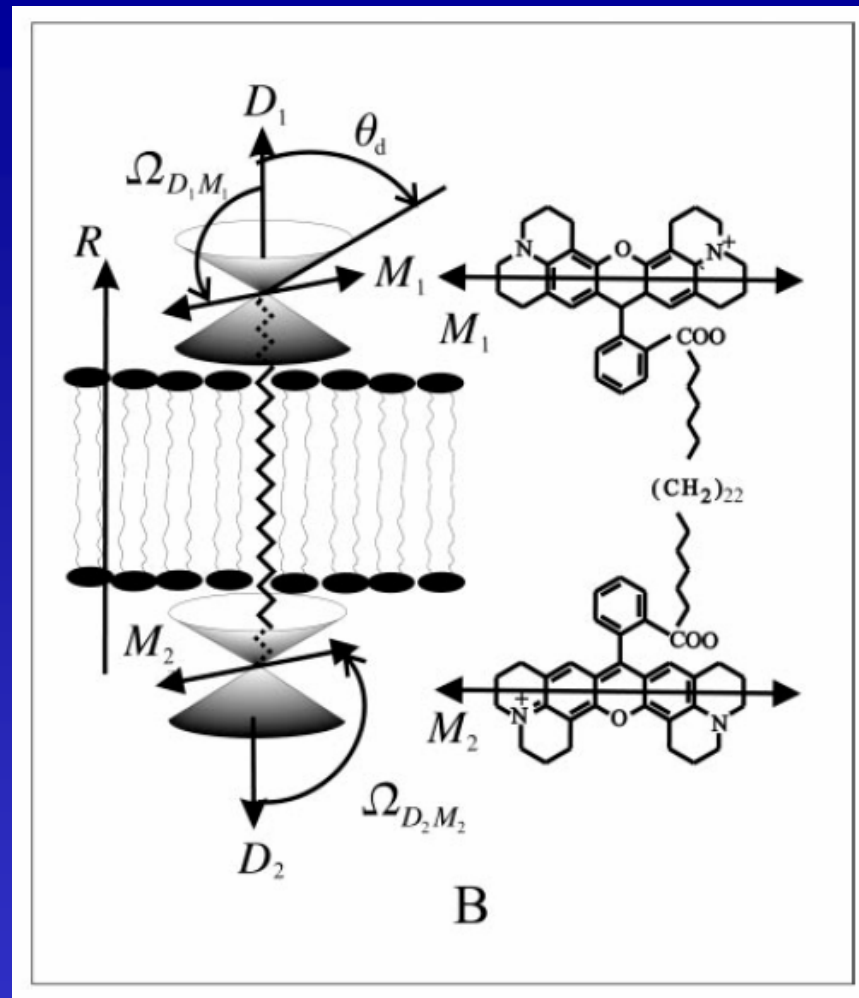
The fluorescence steady-state anisotropy, the initial anisotropy, and the rotational correlation time are denoted by r_s , $r(0)$, and ϕ , respectively. R and R_c denote the distance between the center of mass of the BODIPY groups and the C_α -atoms of the mutated amino acids obtained from the x-ray structure, respectively. The local order of the BODIPY moieties ($j = 1$ and 2) is described by the order parameter S_j , and S_8 accounts for the mutual projection of the symmetry axes of the local orientational distributions. ρ_0 is the maximum contribution to the fluorescence anisotropy from secondary excited donors. The rate of DDEM is ω , and $\langle \kappa^2 \rangle$ accounts for the angular part of dipole-dipole coupling. Further explanation to the notation is given in the text (see *Theoretical Model*) and elsewhere (3, 14). The results represent average values obtained from two to three experiments, using independently prepared samples.

Energy transfer between two identical fluorophores

Phys. Chem. Chem. Phys., 2000, **2**, 2795–2801

$r_s=R_s$: steady-state anisotropy, τ_c =rotational correlation time, D_r =rot. Diffusion constant,

Λ =coupling strength, κ^2 =mean square angular dependence of the dipole-dipole coupling



System	T/K	r_s	$r(0)$	D_r/ns^{-1}	$\tau_{c, \text{theor}}/\text{ns}$	$\tau_{c, \text{exp}}/\text{ns}$
Rh101C ₁₈	293	0.22	0.30 ± 0.02	1.9 ± 0.3	12.0 ± 2.0	7.8 ± 1.5

System	T/K	R_s	$r(0)$	Λ/ns^{-1}	$\langle \kappa^2 \rangle$	$R/\text{\AA}$	$R_{\text{ref}}/\text{\AA}$
Rh101C ₃₂ Rh101	293	0.15	0.28 ± 0.01	5.5 ± 1.5	0.50 ± 0.01	36.5 ± 1.0	38 ^b

Energy transfer between more than two identical fluorophores

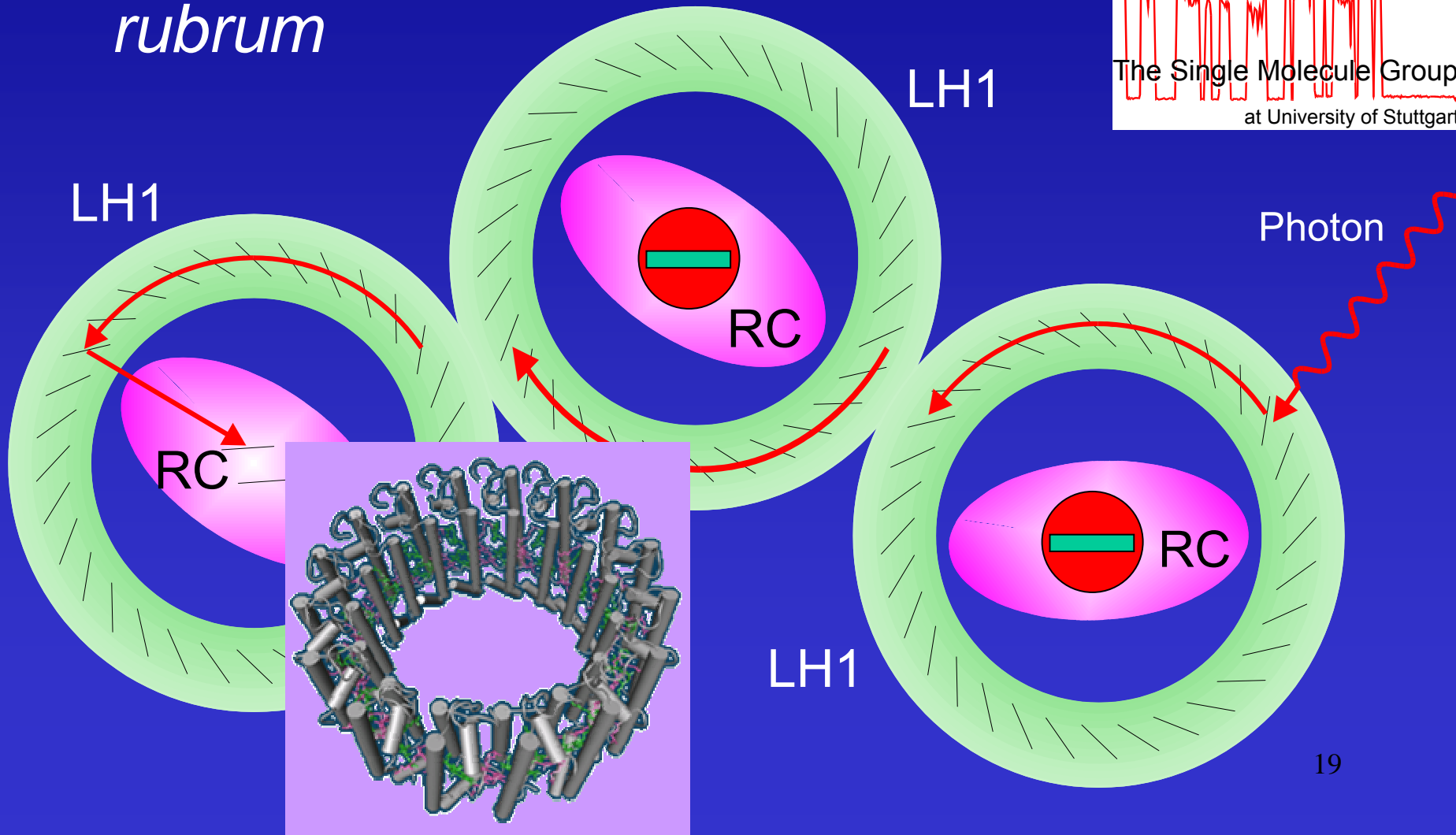
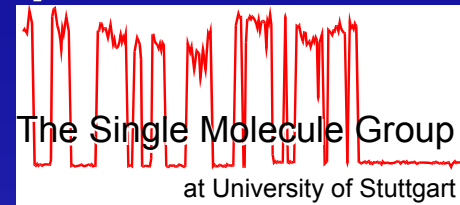
Examples

1. artificial homo-dimers
2. light harvesting complex LH1

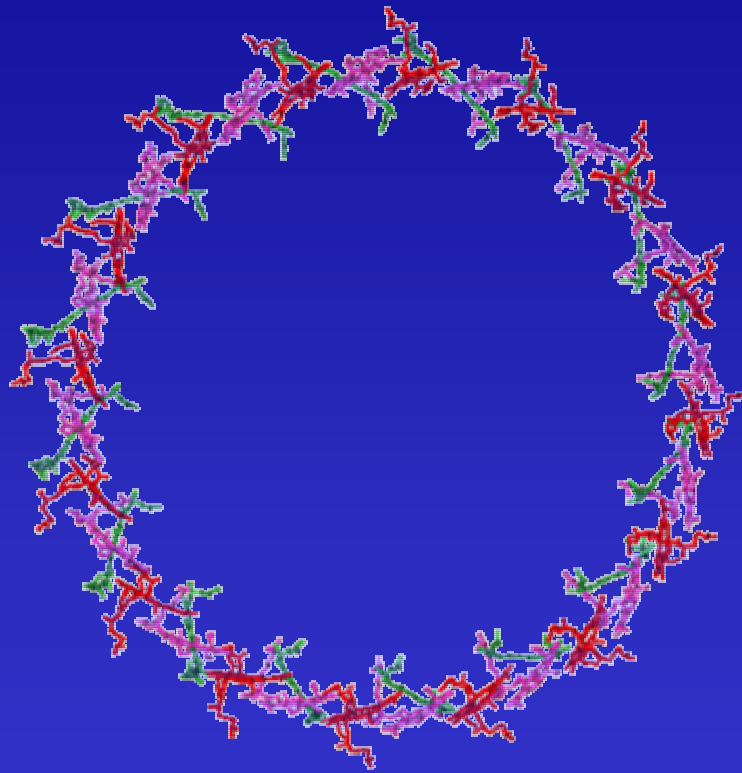
analysis of a multichromophor

the Photosynthetic Unit of *Rhodospirillum*

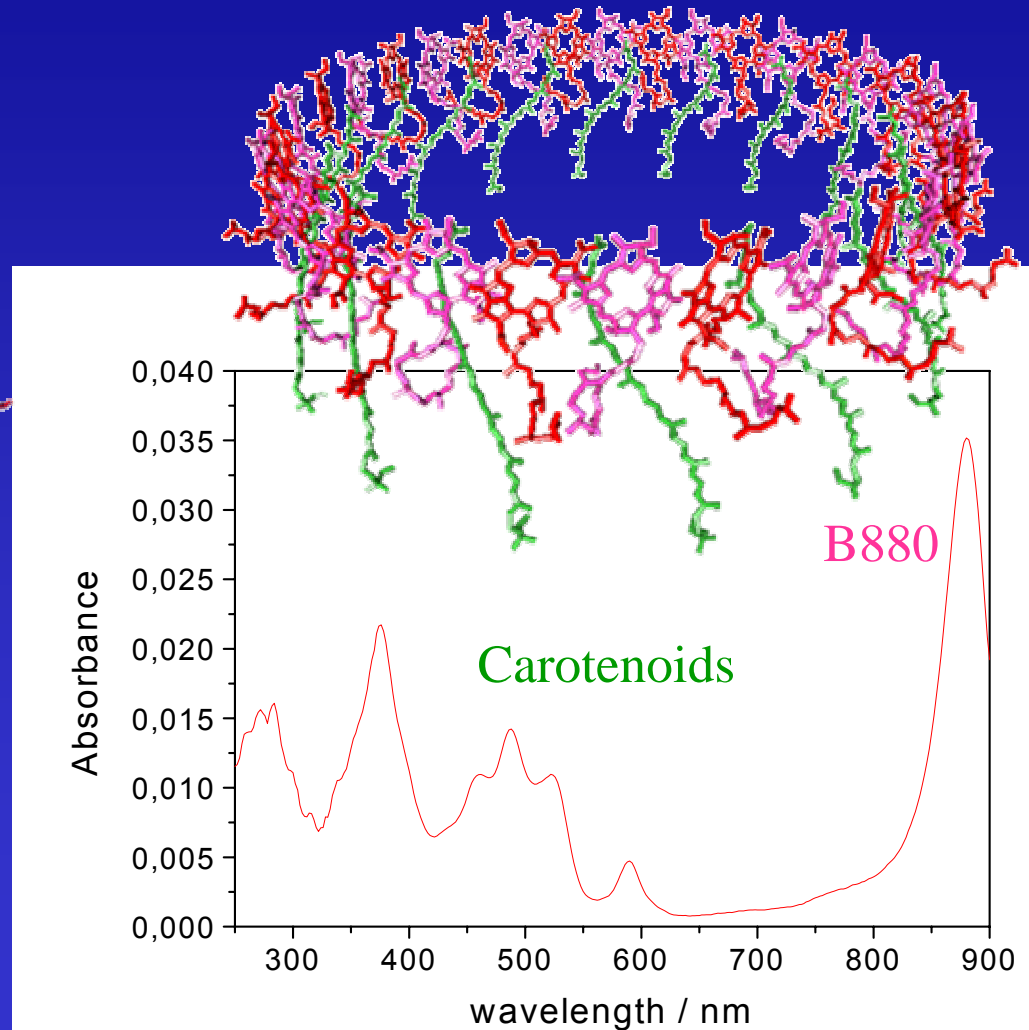
rubrum



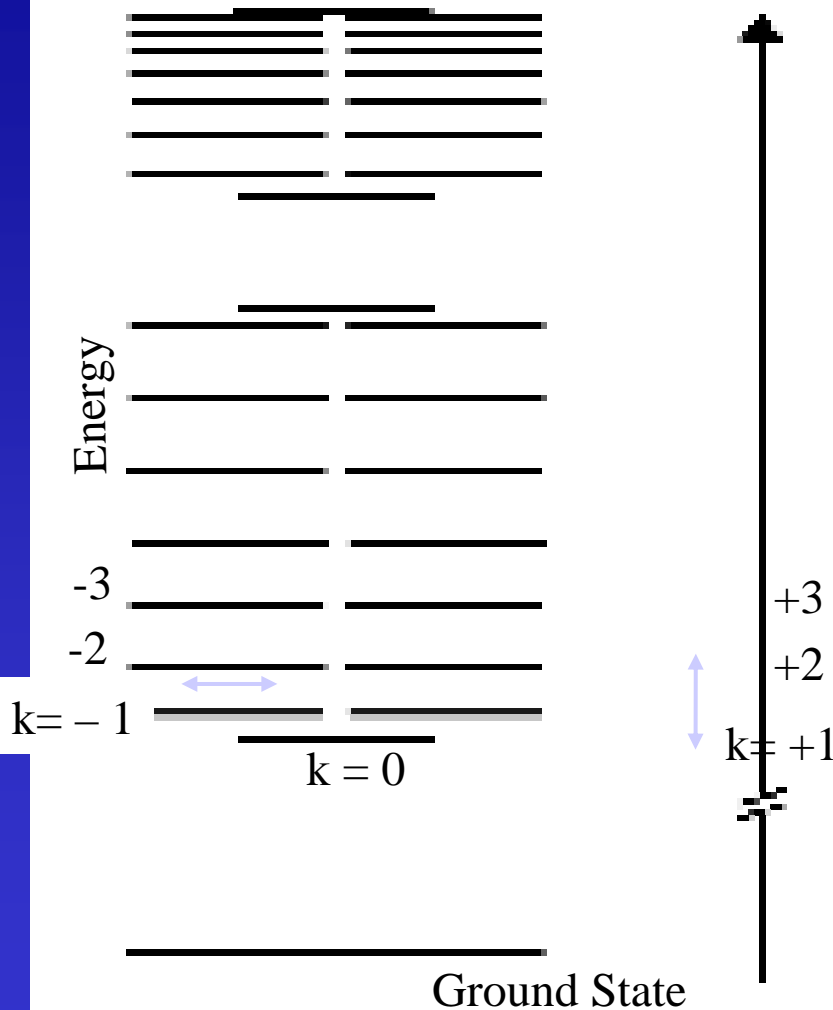
The LH1 structure contains 32 chlorophylls



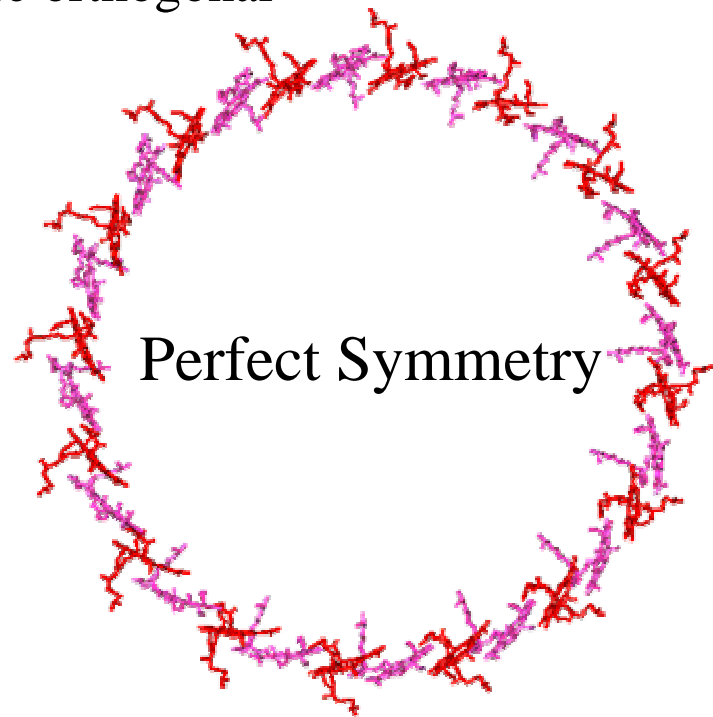
view without proteins



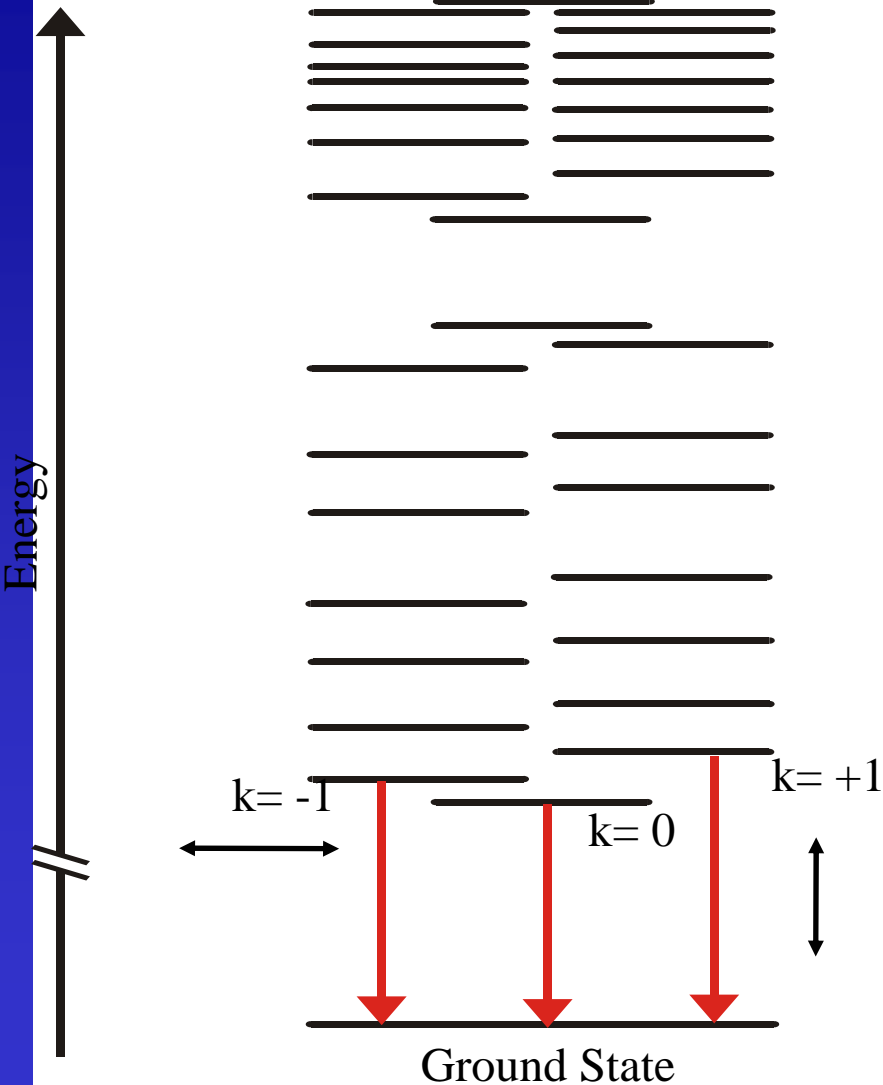
electronic structure of LH1 (I)



- $k=+1$ and $k=-1$ carry almost all oscillator strength
- dipole moments of $k=+1$ and $k=-1$ are orthogonal



electronic structure of LH1 (II)

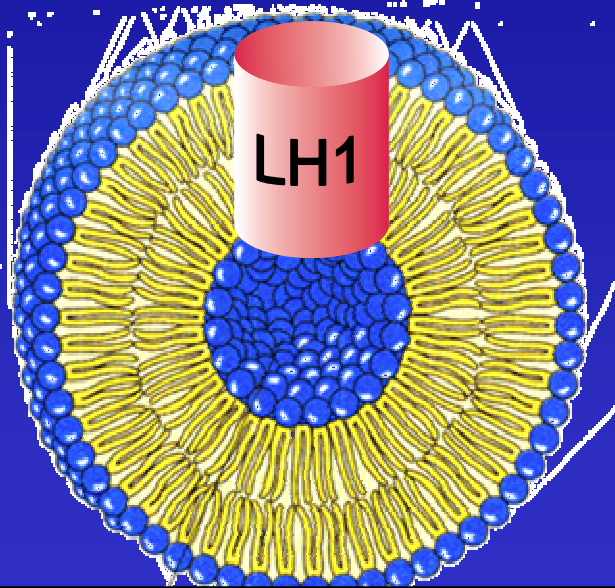


- $k=0$ gets oscillator strength
- **dipole moments of $k=+1$ and $k=-1$ are orthogonal**
- degeneracy is lifted

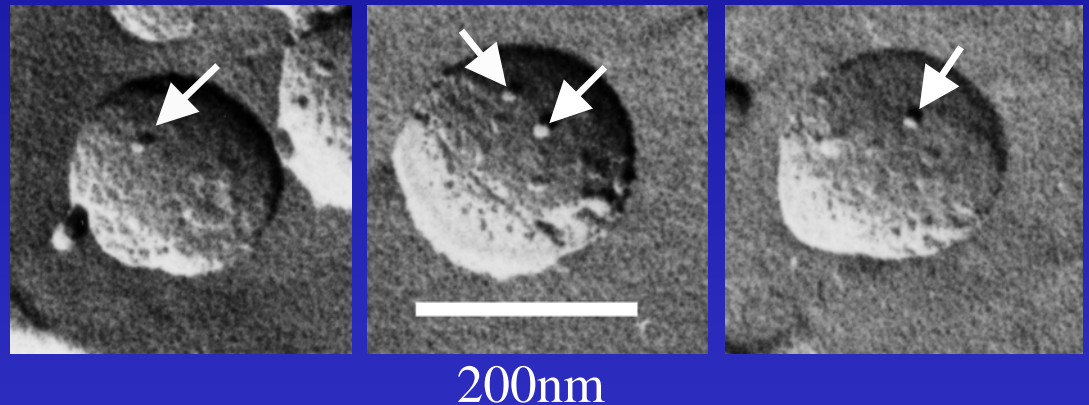
Disordered Ring

LH1 embedded in a lipid vesicle

Freeze-fracture electron micrographs of LH1:

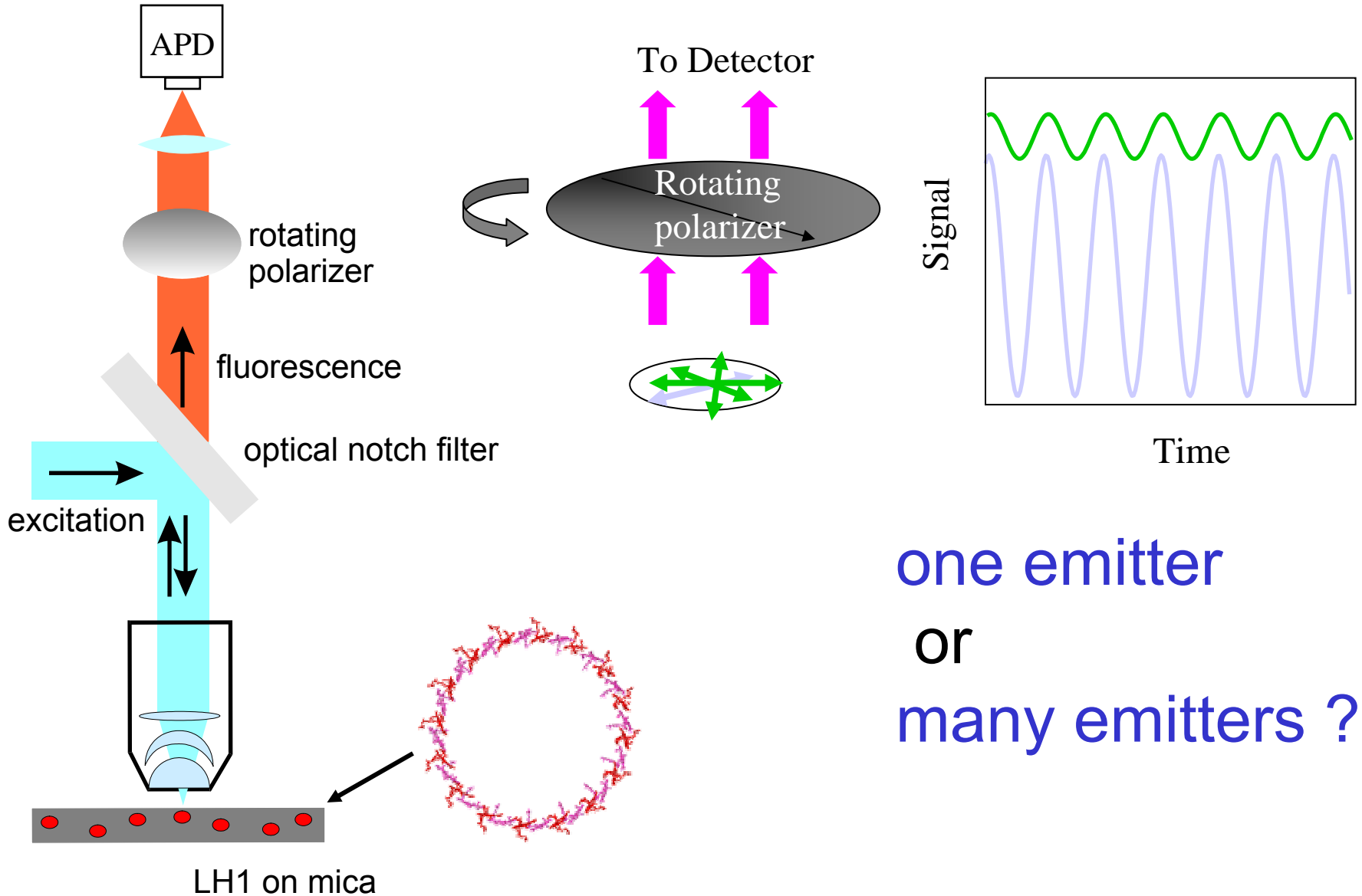


Mica

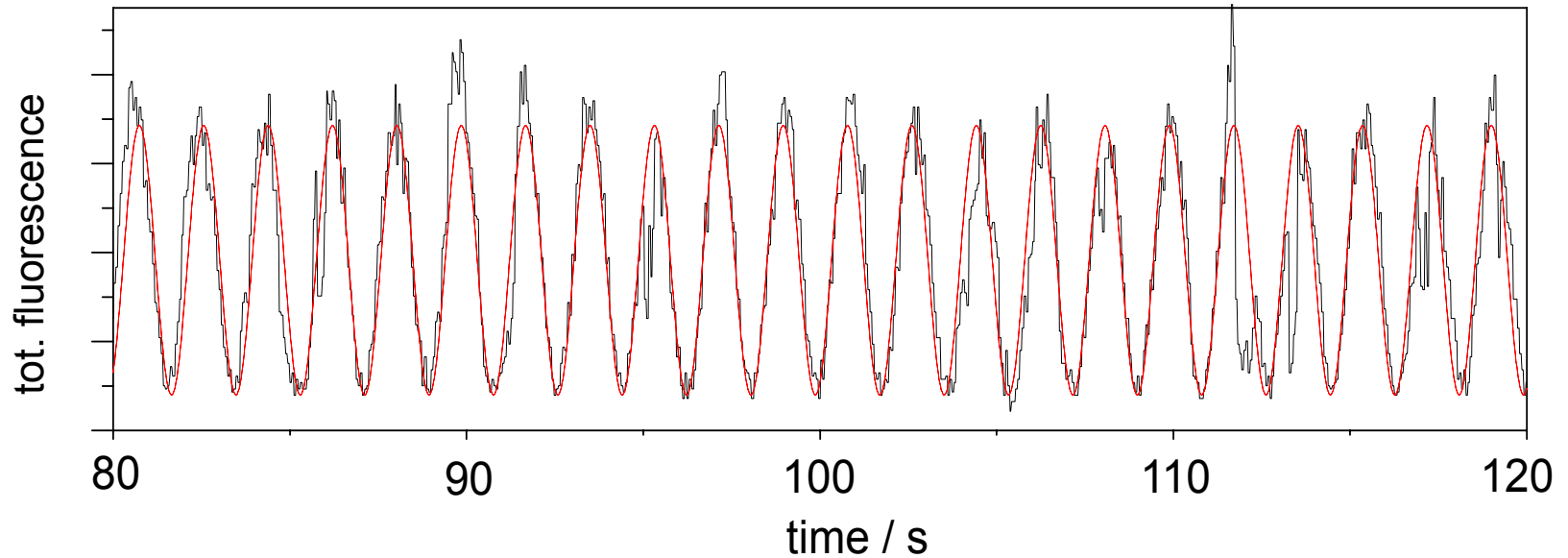


Molar ratio
(LH1 / Phospholipid): 1:400

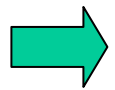
polarization of LH1@ 2K



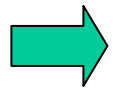
polarization of LH1@ 2K



Fluorescence is linearly polarized.

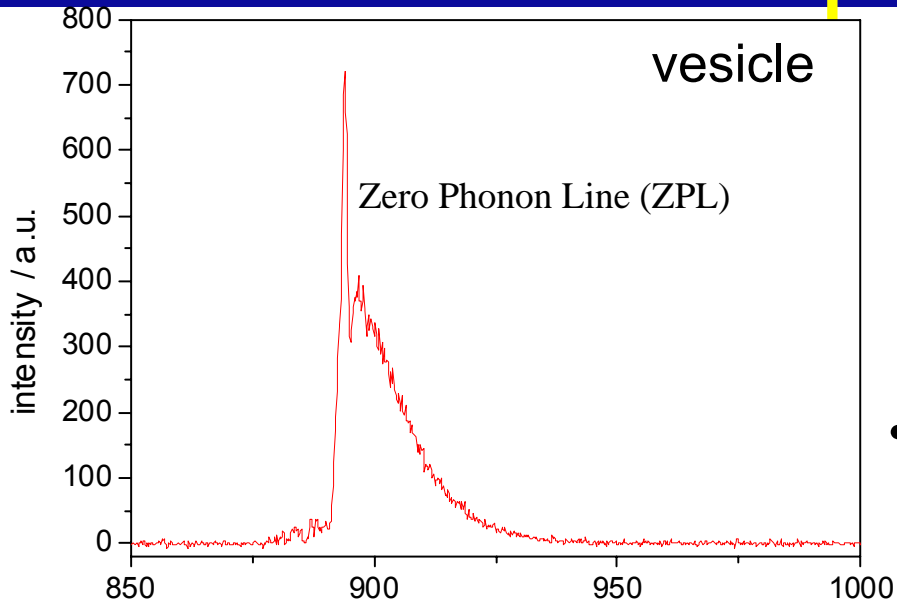


Only one emitting state ($k=0$).

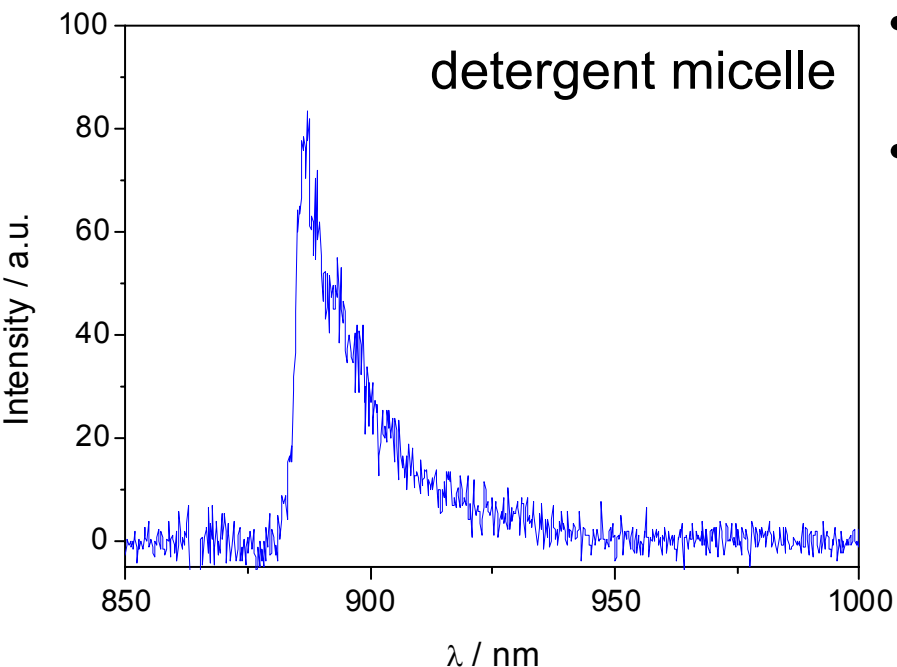


Only one LH1 complex in focus.

fluorescence spectra of LH1 @ 2K

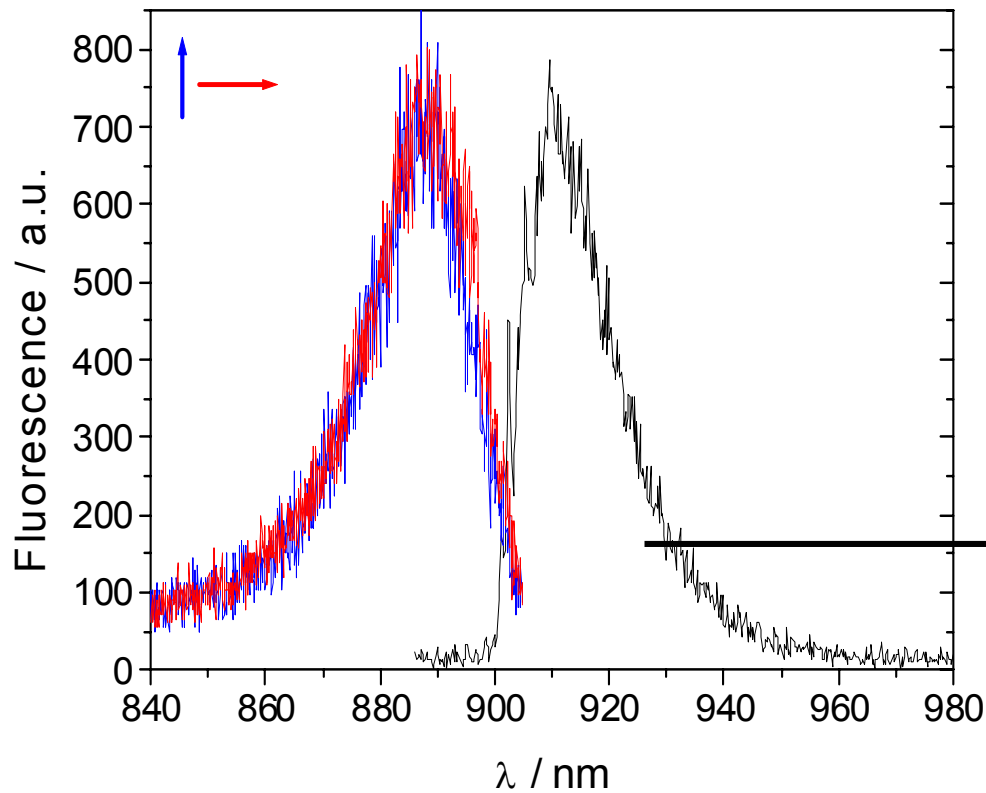


- only lowest transition ($k=0$) is visible in emission



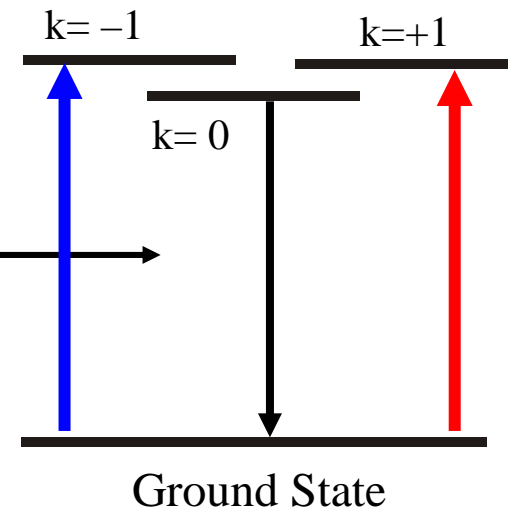
- strong electron-phonon coupling
- spectra of LH1 in membranes are structured due to weaker spectral diffusion

fluorescence excitation spectra @ 2K of LH1 in vesicles



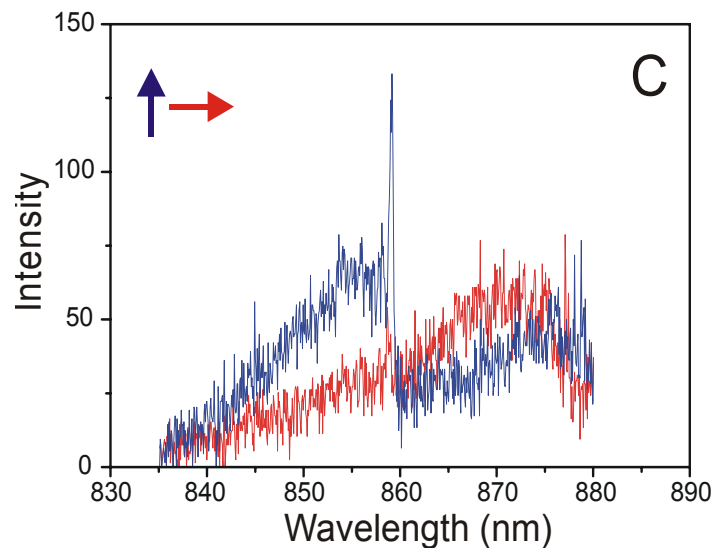
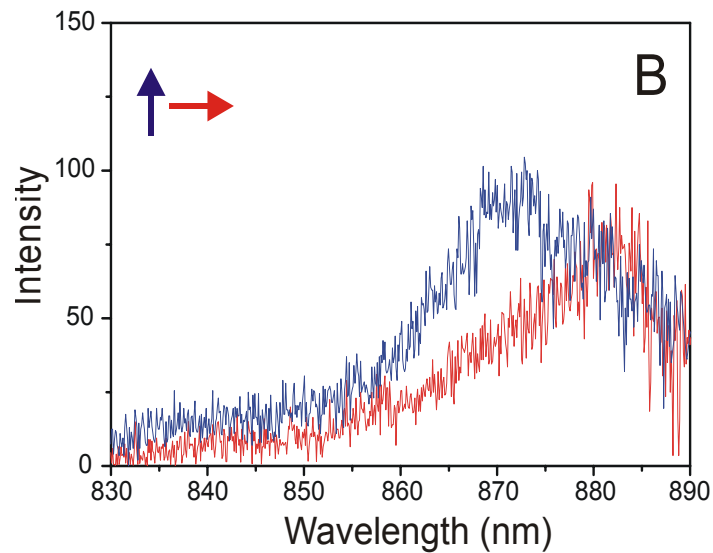
most of the complexes do not
show any splitting

→ nearly circular symmetry



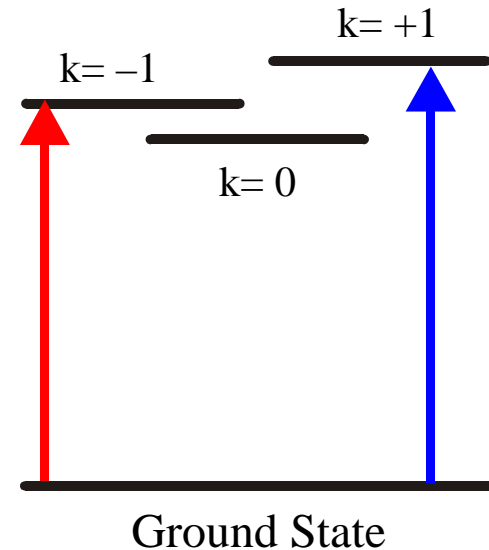
fluorescence excitation spectra

LH1 in detergent



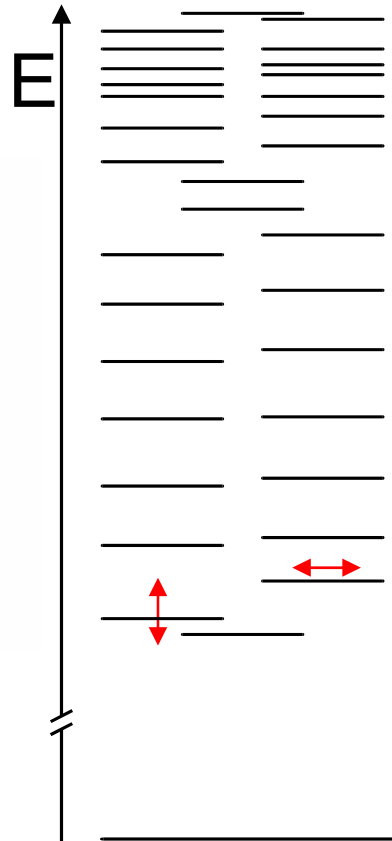
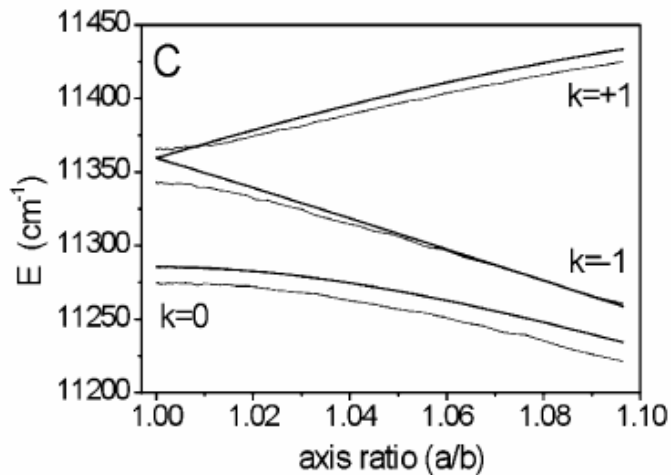
- most of complexes show splitting of $k = \pm 1$ transitions

→ B880 ring is deformed

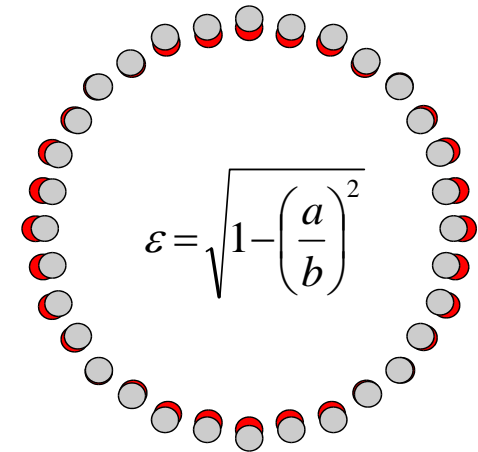


energy level diagram

calculated:



elliptical deformation:
eccentricity: $\varepsilon = 0.3$



splitting of $k = \pm 1$

minor shifting of oscillator strength
to higher lying states

conclusions on LH1

- First single molecule measurements on light-harvesting complexes in a native environment.
- Fluorescence excitation spectroscopy proves nearly undisturbed circular organization of the LH1 complexes in membranes.
- LH1 complexes solubilized with detergent show a wide distribution of elliptically deformations.



The membrane environment is crucial for the structural integrity of the LH1 complex.

U. Gerken, F. Jelezko, C. Tietz, J. Wrachtrup. R. Ghosh, Uni Stuttgart

Biochemistry 2003, 42, 10354–10360

Energy transfer between identical fluorophores

next :

1. artificial homo-dimers
2. light harvesting complexes
3. dendrimers

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